Corrigenda

Cyclisation of 3-(*p*-Methylphenyl)propan-1-ol *via* its Alkoxyl Radical and Aryl Radical Cation Intermediates. A Comparison of Regioselectivities

André Goosen, Cedric W. McCleland and Fabrizio C. Rinaldi

J. Chem. Soc., Perkin Trans. 2, 1993, 279

One of the results reported in the above paper is unfortunately incorrect.

| The sixth entry in Table 1 which currently reads | | | | |
|---|-----|-----|------|-----------|
| 1.4(3.2) | 2.6 | 0.2 | 56.9 | 7.1:92.9 |
| should read | | | | |
| 1.4(3.2) | 1.3 | 1.5 | 56.9 | 53.6:46.4 |

Fig. 1 is consequently affected and should be redrawn as shown.



Fig. 1 Effect of pH on chroman ratio: ■, radical cation 6; ▲, alkoxyl radical 7

Furthermore, the sentence 'However, in contrast . . . at low pH', which appears in the penultimate paragraph of the discussion, should now read:

'However, in contrast to the alkoxyl radical reaction where the 7-methylchroman 9 remains the major component over the entire pH range, the proportion of chroman 9 resulting from the radical cation is significantly reduced at lower pH.'

Formation of Cyclopropylsulfones from 1-Arylsulfonyl-2-chloromethylprop-2-enes

Stephen M. Jeffery and Charles J. M. Stirling

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